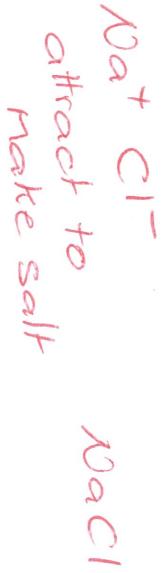


Types of bonds (define and diagram)

Ionic bond - Occurs between ions + and -

electrical charge



Covalent bond - atoms share electrons

Sharing is equal - non-polar
or
not equal share - partially
negative charge + lesser
share partially + charge
- polar



O - partially negative
H - partially positive



Hydrogen bond -

formed by attraction of

Polar molecules

Partially + → Partially -
attract

Not very strong

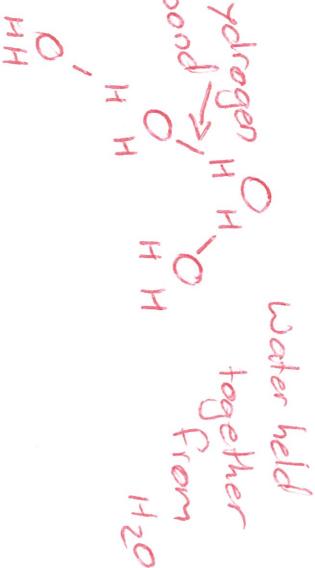
Water held

hydrogen bond → H-H O-O H-H

together from

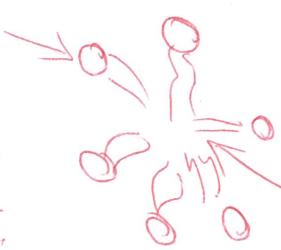
H₂O

O'



Hydrophilic - Likes water - will dissolve in

(Polar) hydrophilic



form hydrogen bonds
with H₂O - so they
dissolve

cell
membranes

hydrophobic

Hydrophobic - Doesn't like H₂O - clumps together to get away
from H₂O

will dissolve

in another

hydrophobic

substance

no hydrogen bonds

- Don't dissolve in

H₂O

Synthesising biomacromolecules

- Autotroph –

- Chemotroph –

- Heterotroph –

Match the definition

Chem

• an organism that synthesis food through chemical processes other than photosynthesis.

Hetero

organisms that consume other organisms as food. They are not able to make organic molecules from simple inorganic compounds.

Auto

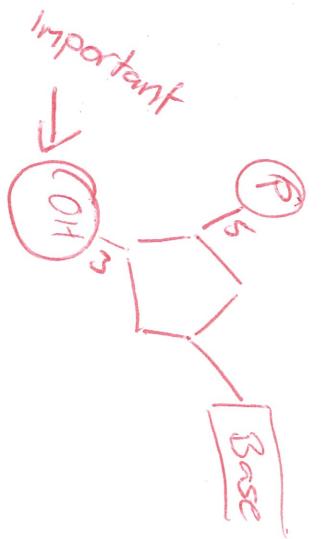
an organism that makes its own food from light energy or chemical energy without eating.

Polymerisations — joining of monomers — form long chain
 link small repeating molecules

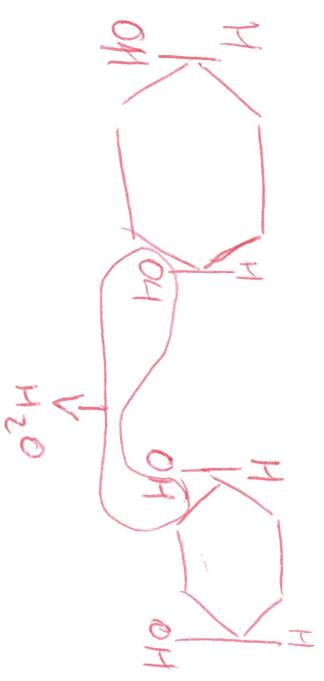
↓
 link small repeating molecules

Example

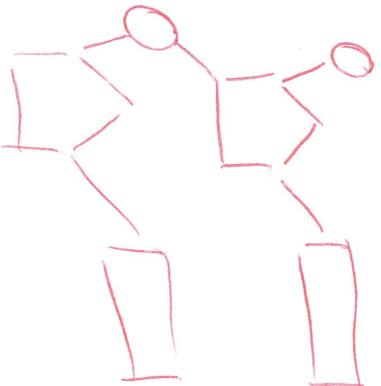
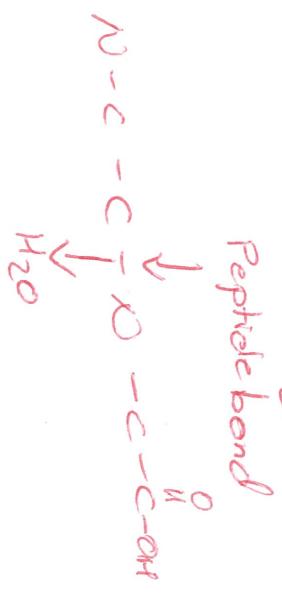
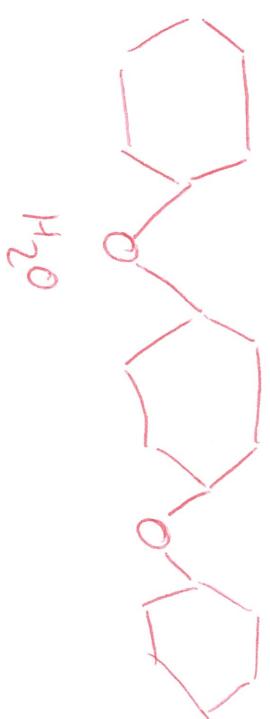
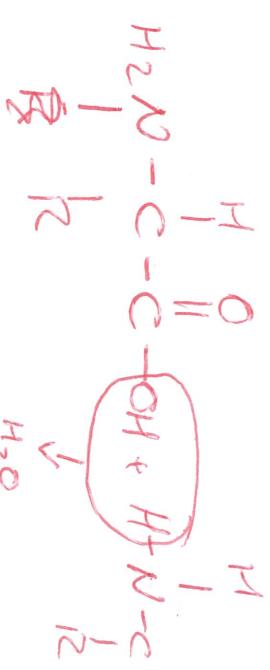
Nucleic acid



carbohydrates



Proteins



Nucleic acids

~~POLYMER~~

Sugar
Phosphate
Nitrogenous base

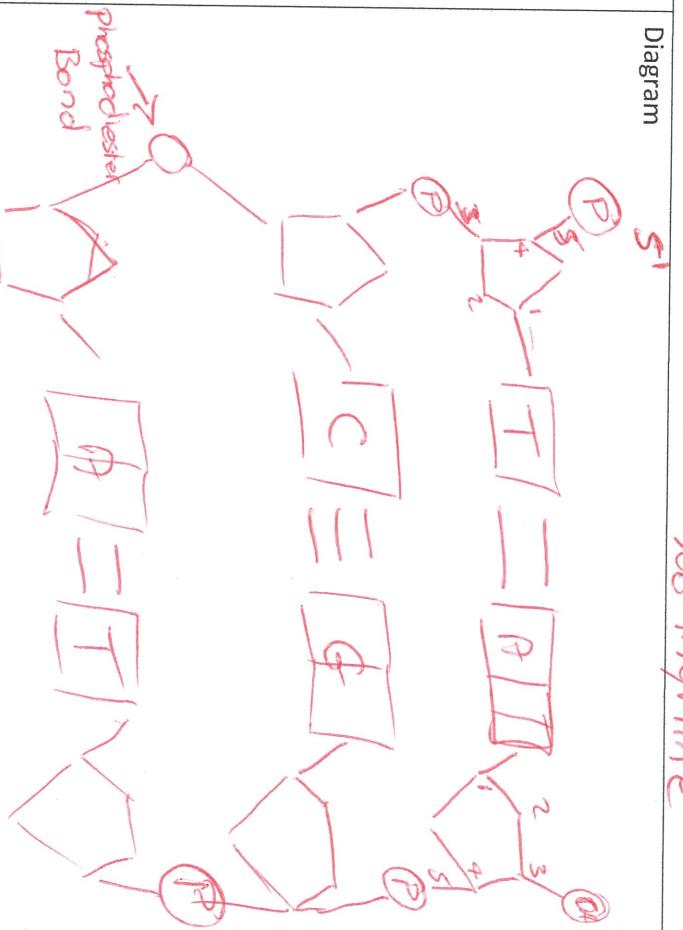
RNA - Uracil
No Thymine

C = G
A = T

Facts

- * Found nucleus + Mitochondria
- * DNA code
 - what to do
 - what proteins to make
- * Double Helix
- * RNA single strand
- * Link together 5' to 3'
 - Used in DNA replication
 - + translation for RNA
- * 3' hydroxyl group - important for
 - put into codons for protein synthesis
 - code for amino acids
- Summary sentence
 - genes code for proteins - only ~10% DNA actually codes for proteins
- * codon table - amino acid code
- * RNA take DNA code to mRNA (translation)
 - Ribosome - Ribosomes read mRNA
 - tRNA bring codon to make protein (transcription)

Diagram



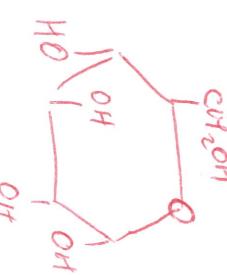
Carbohydrates



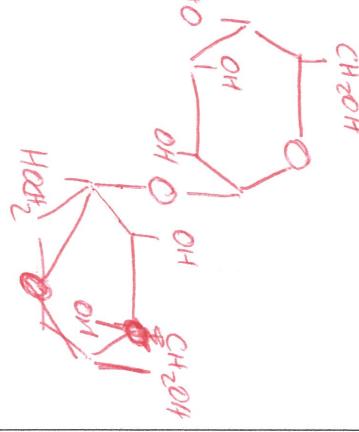
Facts

- * Source energy and for structural component
- * - makes glycoproteins
- * cellulose - structural support to plants
 - no nutrient value to humans
 - bacteria helps animals break it down

Diagram



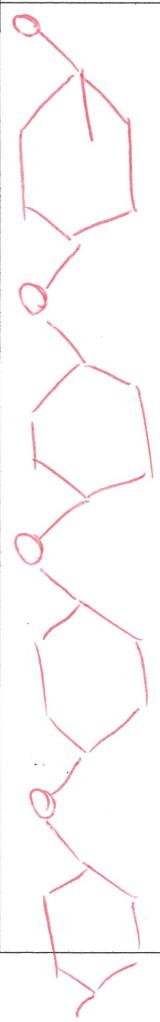
disaccharide (sucrose)



Triose
Pentose
Hexose

hectose
matrose

Polysaccharide (starch)



Summary sentence

Lipids

H, C, O, P, N

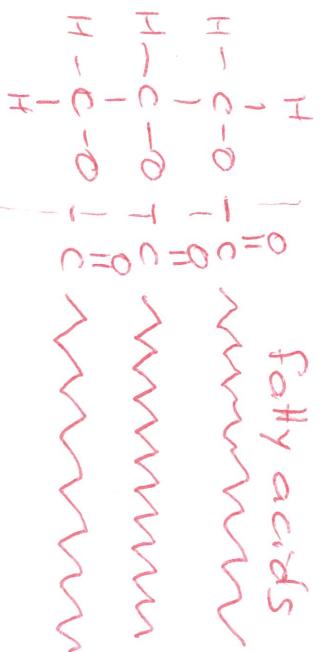
Phospholipid
hydrophilic Polar
hydrophobic non-polar

Facts

- * Fats, oils + waxes
- * hydrophobic, some hydrophilic, some both (ends)
- * Function
 - energy storage
 - structural component membrane
 - specific biological function

Diagram

Glycerol



fatty acids

Saturated
found in animals

solid

Unsaturated
found in plants

Liquid

c=c

Max number H bonded
to each carbon

Double bonds before

Some C
not max number
H

Summary sentence

Protein

N, C, H, O

Facts

- Proteome - whole set of proteins produced by a cell
- involved in structure of cells + many chemical reactions - using enzymes
- Made 20 amino acids
- Amino acids bond to form linear polypeptides - fold, twist + coil
- Arrangement of folds give proteins different characteristics.

* can denature

- small can return to normal

Diagram

Primary structure Amino end
Dna determines $\text{N}-\text{C}-\text{C}-\text{N}-\text{C}-\text{C}-\text{N}-\text{C}$
Sequence |
 |
 |
 R

carboxyl end

Secondary
 α helix - hydrogen bonds between amino acids at different locations

peptides, strong
peptide bonds

β Pleated sheet



Quaternary

more than 1 polypeptide

e.g. Haemoglobin

Tertiary
Many loops + folds
Gives them shape



Quaternary

more than 1 polypeptide

e.g. Haemoglobin

Summary sentence

- Types**
- Motility
 - Structural
 - Enzymes
 - transport
 - Hormones
 - cell-surface receptors
 - Neurotransmitters
 - Immunoglobulins
 - Poisons + toxins

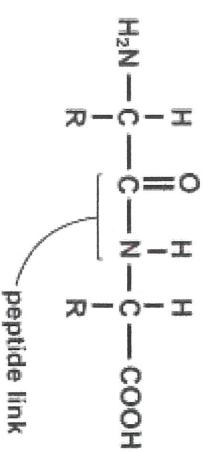
What reaction is this?

Condensation Reaction

- make water



water lost

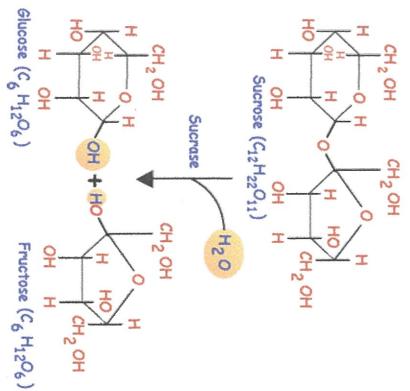


peptide link

What is the reverse called?

hydrolysis Reaction

- add water to separate into monomers (smaller pieces)



Glucose ($C_6H_{12}O_6$)

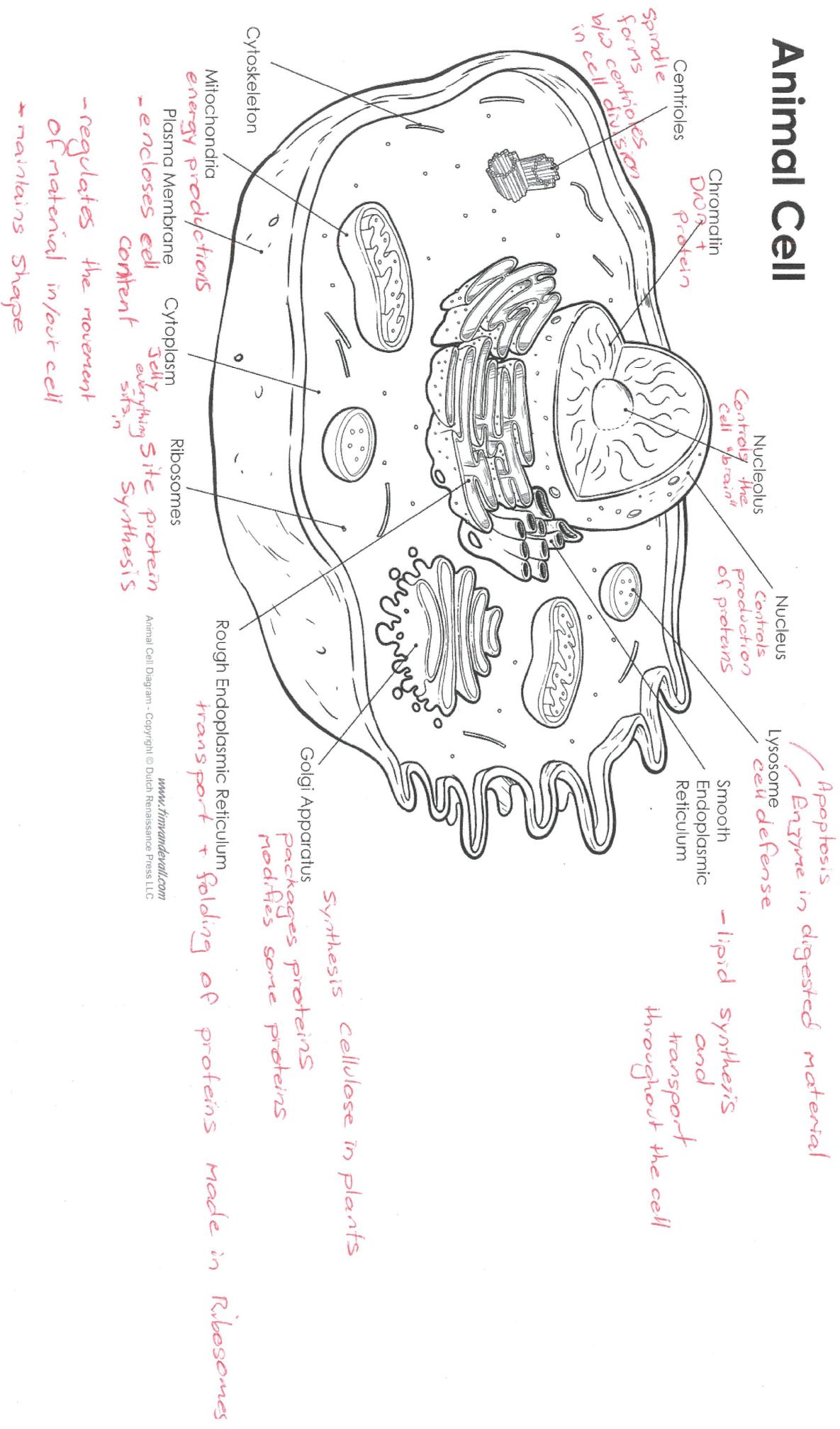
Fructose ($C_6H_{12}O_6$)

Types cells



What's my job in the cell? (tell me what each organelle in the cells job is)

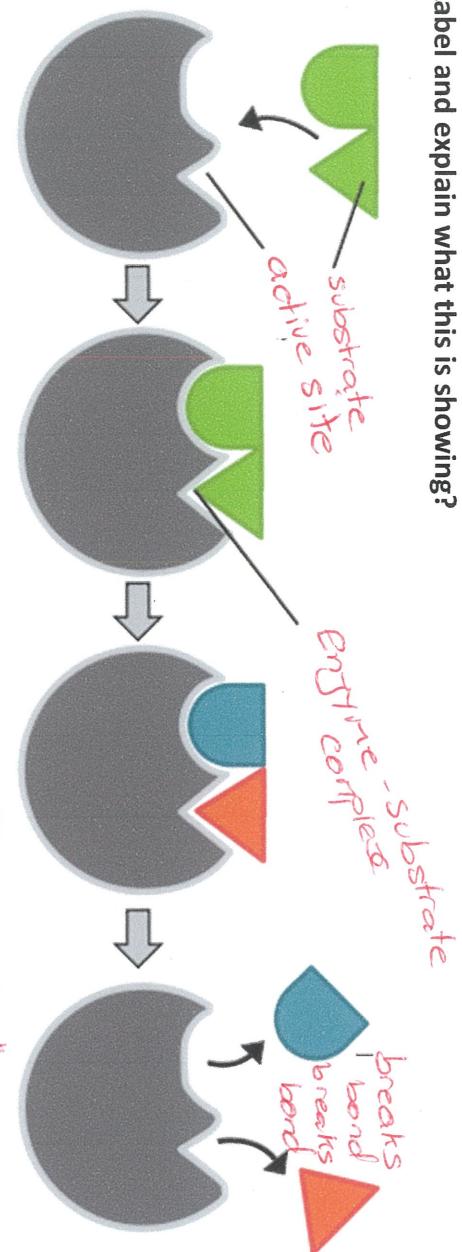
Animal Cell



What type of cell am I found in?

Organelle	Prokaryotic	Animal cell	Plant cell
Cell membrane	✓	✓	✓
Nucleus		✓	✓
Nuclear membrane		✓	✓
Mitochondria		✓	✓
Ribosomes	✓	✓	✓
Smooth endoplasmic reticulum		✓	✓
Rough endoplasmic reticulum		✓	✓
Golgi complex		✓	✓
Lysosome		✓	✓
Vacuole		✓	✓
Chloroplast			✓ some
Centriole		✓	✓ some
Flagella		✓ some	✓ some
Cell wall	✓		✓ some

Label and explain what this is showing?



What would denaturing do to this?



Induced - Fit model
— enzyme + substrate
can modify the
shape of the
enzyme to accomodate

Energy usually
released in these
reactions

changes shape of active site
no longer binding site

What causes denaturing to occur?

Heat
pH

What are the 2 ways inhibitors work?

Competitive

- binds active site
- temporary
- prevents binding of enzyme + substrate

Non-competitive

- binds with enzyme at another part of molecule
- Alters shape of enzyme
- can't bind to active site

How to we develop drugs that effect the function of an enzyme?

each enzyme unique fit for a function

- Develop drugs that inhibit the activity of enzymes - competitive inhibitor

Binding Poisons - stop function of enzyme

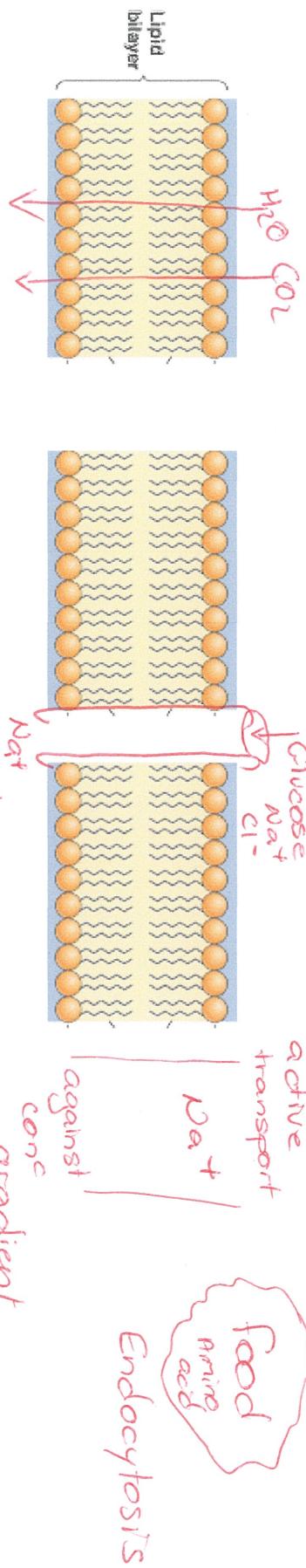
Co-enzymes

- chemical component - add to enzyme to be active

For the substances listed below draw in the cell membrane what they require in order to cross the membrane and show how they cross the membrane.

Water, carbon dioxide, glucose, chloride, large food particle, polar molecule, amino acids.

This cell has a high concentration of sodium inside the cell but still requires more in order to function well, how will it get into the cell?

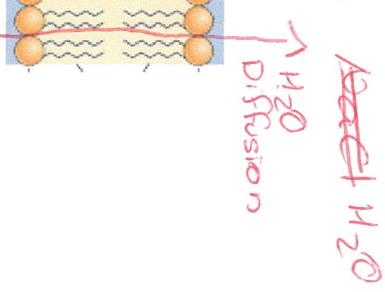
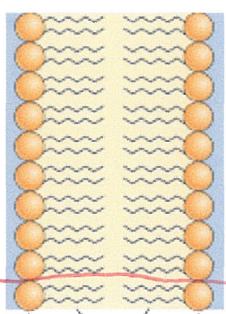
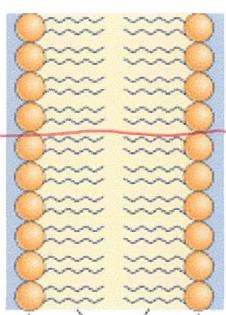
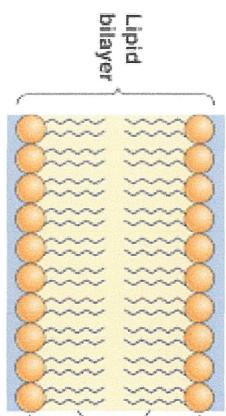


Use the diagram above to show the following terms

Diffusion *Through phospholipids*

Facilitated diffusion *Protein channel*

Active transport *-against conc gradient*

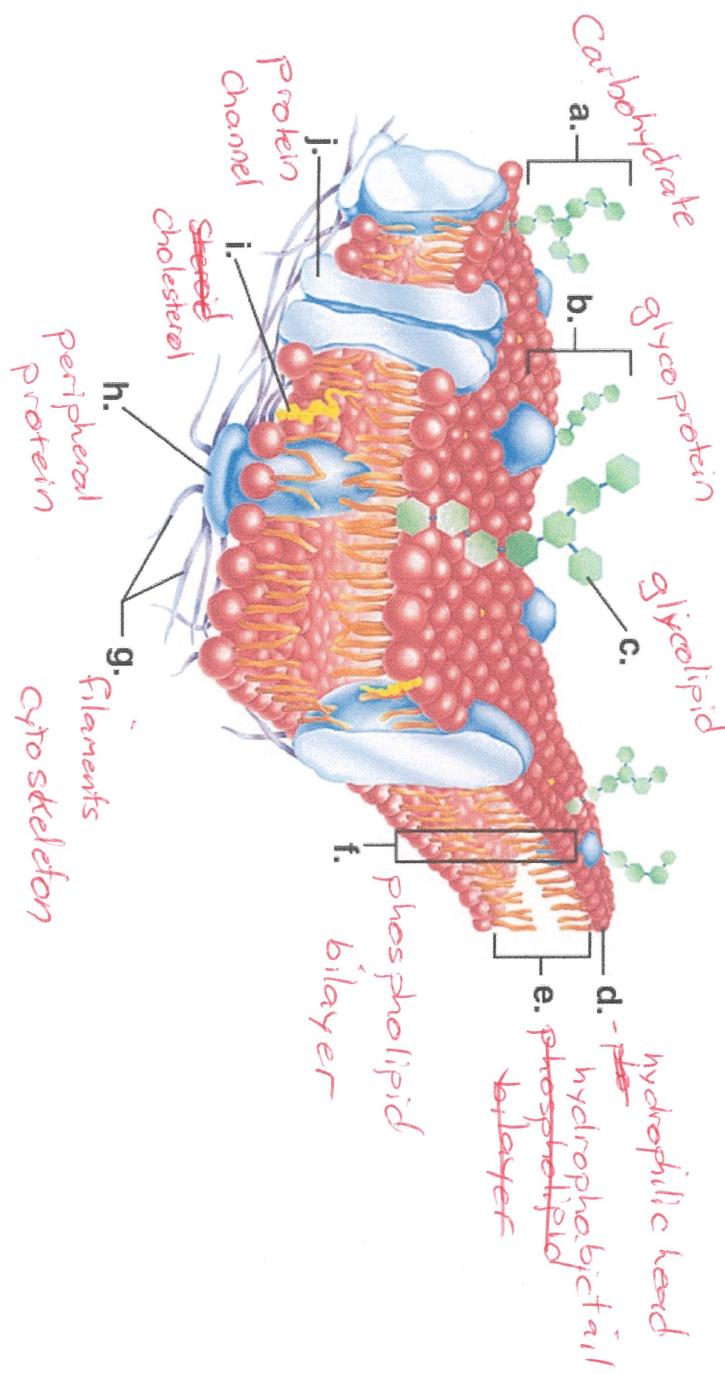


$Hi \cancel{H_2O} H_2O$

Explain the difference between endocytosis and exocytosis. Use a diagram to help with this explanation.

Label

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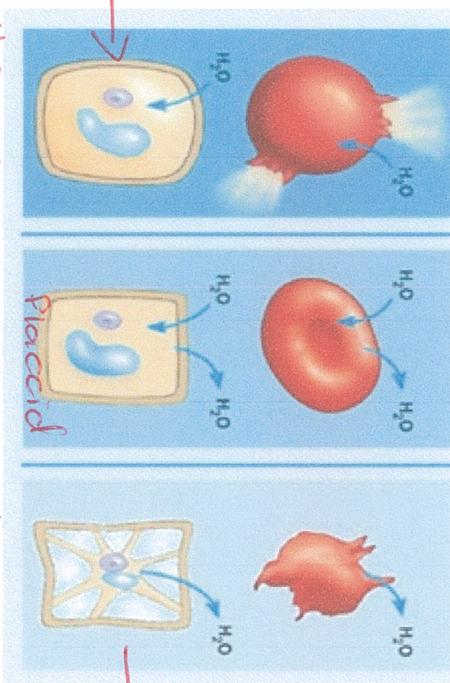


What is the difference between osmosis and diffusion?

1. H₂O moves across membrane and disperses randomly into available space
Solid, liquid, gas randomly available
high conc to low conc

Name and explain the following diagrams

Hypotonic solution Isotonic solution Hypertonic solution



Reactions for cells

H₂O move into cell
Same conc
no net movement

Why do cells need energy?

Plasmolysis

What is another name for an anabolic reaction?

What is an anabolic reaction?

What is another name for a catabolic reaction?

What is a catabolic reaction?

What is ATP?

What is the chemical equation for ATP?

$\text{ATP} \rightarrow$

What is the equation of respiration?

What is the equation for photosynthesis?

Explain the process of respiration

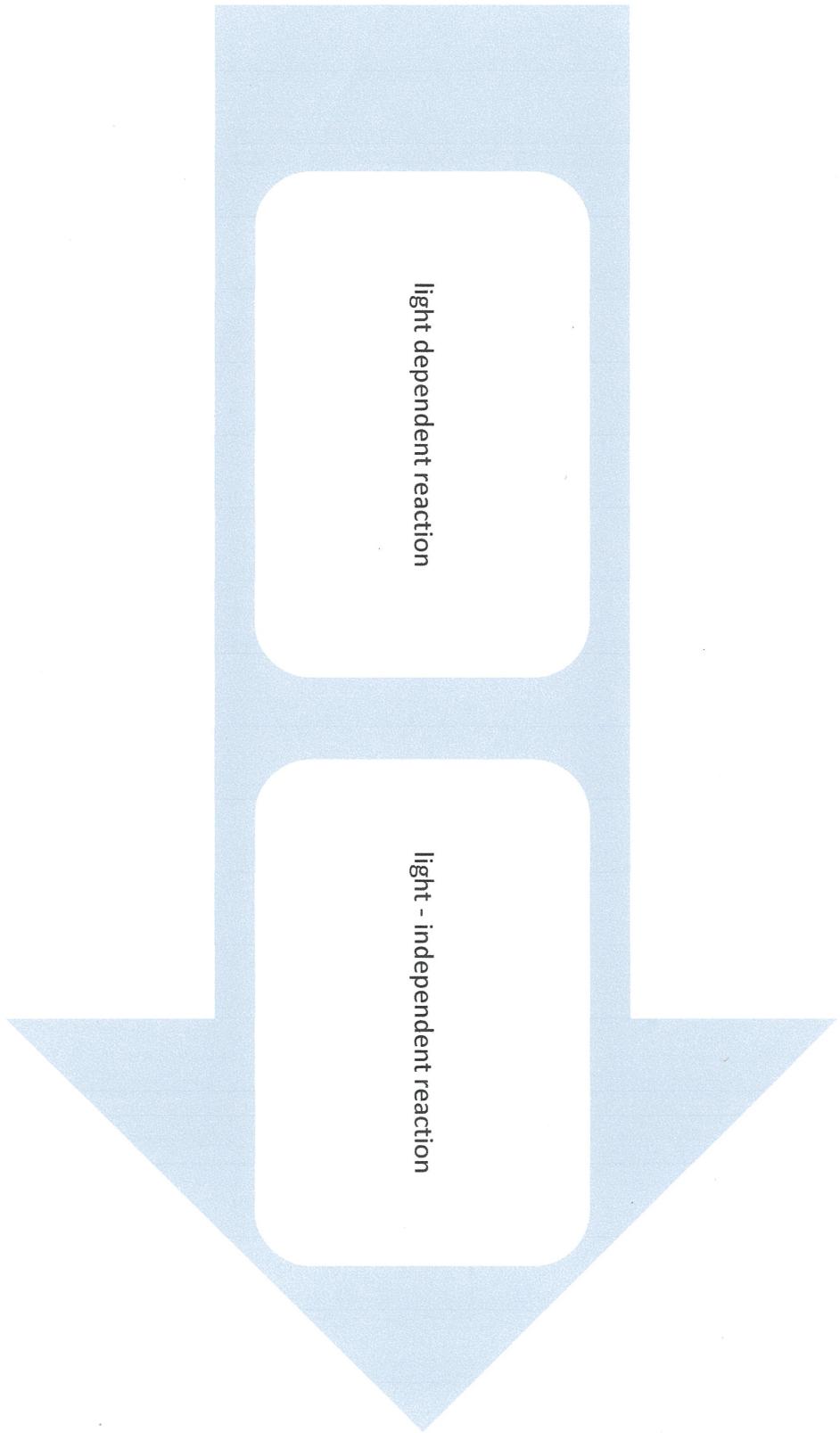
glycolysis

kreb cycle

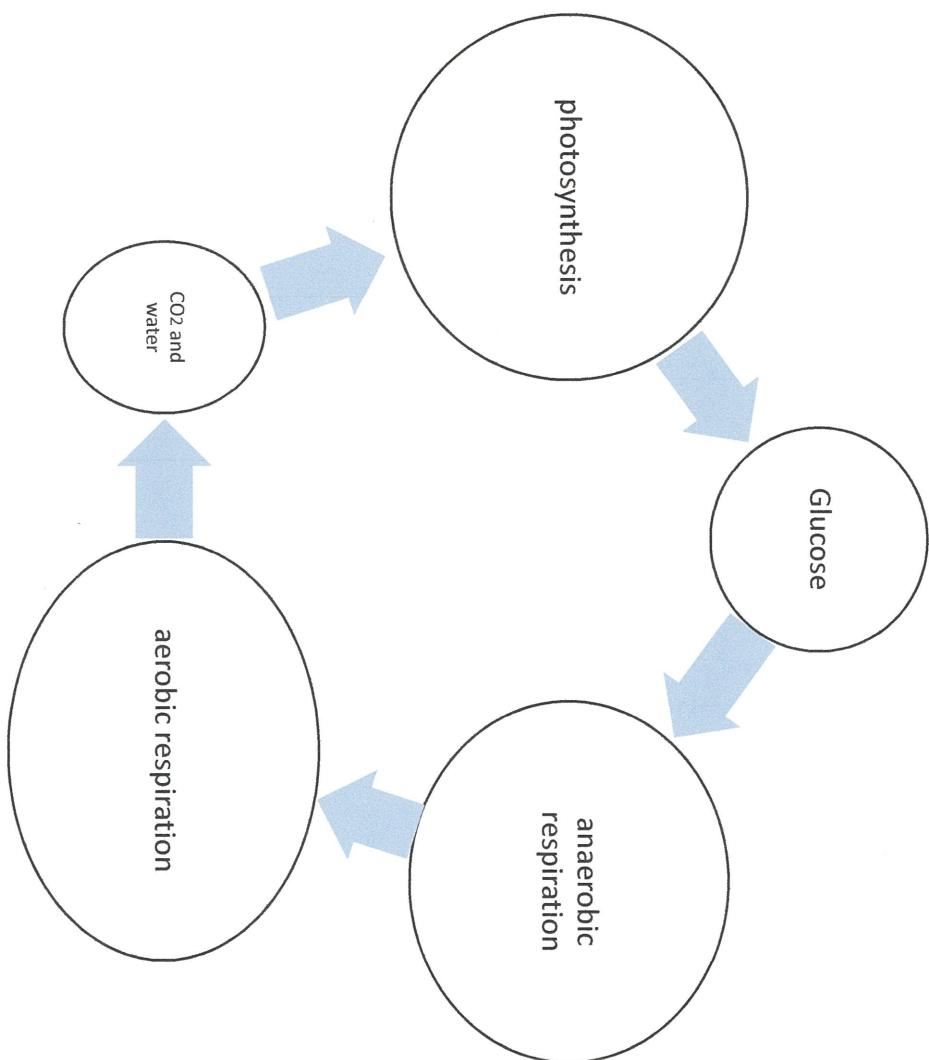
electrontransport

How does aerobic respiration differ to anaerobic respiration?

Explain the photosynthetic process



Add detail to the cycle below to explain what is happening



To the cycle above add in the ATP – ADP cycle